# SOLUTIONS MANUAL TO ACCOMPANY INTRODUCTION TO FLIGHT $8^{\text {th }}$ Edition 

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## Chapter 2

$2.1 \quad \rho=p / R T=(1.2)\left(1.01 \times 10^{5}\right) /(287)(300)$
$\rho=1.41 \mathrm{~kg} / \mathrm{m}^{2}$
$v=1 / \rho=1 / 1.41=0.71 \mathrm{~m}^{3} / \mathrm{kg}$
2.2 Mean kinetic energy of each atom $=\frac{3}{2} k T=\frac{3}{2}\left(1.38 \times 10^{-23}\right)(500)=1.035 \times 10^{-20} \mathrm{~J}$

One kg-mole, which has a mass of 4 kg , has $6.02 \times 10^{26}$ atoms. Hence 1 kg has $\frac{1}{4}\left(6.02 \times 10^{26}\right)=1.505 \times 10^{26}$ atoms.

Total internal energy $=($ energy per atom)(number of atoms)

$$
=\left(1.035 \times 10^{-20}\right)\left(1.505 \times 10^{26}\right)=1.558 \times 10^{6} \mathrm{~J}
$$

$2.3 \quad \rho=\frac{p}{R T}=\frac{2116}{(1716)(460+59)}=0.00237 \frac{\text { slug }}{\mathrm{ft}^{3}}$
Volume of the room $=(20)(15)(8)=2400 \mathrm{ft}^{3}$
Total mass in the room $=(2400)(0.00237)=5.688$ slug

$$
\text { Weight }=(5.688)(32.2)=183 \mathrm{lb}
$$

$2.4 \quad \rho=\frac{p}{R T}=\frac{2116}{(1716)(460-10)}=0.00274 \frac{\text { slug }}{\mathrm{ft}^{3}}$
Since the volume of the room is the same, we can simply compare densities between the two problems.
$\Delta \rho=0.00274-0.00237=0.00037 \frac{\text { slug }}{\mathrm{ft}^{3}}$
$\%$ change $=\frac{\Delta \rho}{\rho}=\frac{0.00037}{0.00237} \times(100)=15.6 \%$ increase
2.5 First, calculate the density from the known mass and volume, $\rho=1500 / 900=1.67 \mathrm{lb}_{\mathrm{m}} / \mathrm{ft}^{3}$ In consistent units, $\rho=1.67 / 32.2=0.052$ slug $/ \mathrm{ft}^{3}$. Also, $T=70 F=70+460=530 R$.

Hence,

$$
\begin{aligned}
p & =\rho R T=(0.52)(1716)(530) \\
\text { or } \quad p & =47,290 \mathrm{lb} / \mathrm{ft}^{2} \\
p & =47,290 / 2116=22.3 \mathrm{~atm}
\end{aligned}
$$

2.6
$p=\rho R T$
$\ell n p=\ell n p+\ell n R+\ell n T$
Differentiating with respect to time,
$\frac{1}{p} \frac{d p}{d t}=\frac{1}{\rho} \frac{d \rho}{d t}+\frac{1}{T} \frac{d T}{d t}$
or, $\frac{d p}{d t}=\frac{p}{\rho} \frac{d \rho}{d t}+\frac{p}{\mathrm{~T}} \frac{d T}{d t}$
or, $\quad \frac{d p}{d t}=R T \frac{d \rho}{d t}+\rho R \frac{d T}{d t}$
At the instant there is $1000 \mathrm{lb}_{\mathrm{m}}$ of air in the tank, the density is

$$
\begin{aligned}
& \rho=1000 / 900=1.11 \mathrm{lb}_{\mathrm{m}} / \mathrm{ft}^{3} \\
& \rho=1.11 / 32.2=0.0345 \mathrm{slug} / \mathrm{ft}^{3}
\end{aligned}
$$

Also, in consistent units, is given that

$$
T=50+460=510 R
$$

and that

$$
\frac{d T}{d t}=1 F / \mathrm{min}=1 R / \mathrm{min}=0.016 R / \mathrm{sec}
$$

From the given pumping rate, and the fact that the volume of the tank is $900 \mathrm{ft}^{3}$, we also have

$$
\begin{aligned}
& \frac{d \rho}{d t}=\frac{0.5 \mathrm{lb}_{\mathrm{m}} / \mathrm{sec}}{900 \mathrm{ft}^{3}}=0.000556 \mathrm{lb}_{\mathrm{m}} /\left(\mathrm{ft}^{3}\right)(\mathrm{sec}) \\
& \frac{d \rho}{d t}=\frac{0.000556}{32.2}=1.73 \times 10^{-5} \mathrm{slug} /\left(\mathrm{ft}^{3}\right)(\mathrm{sec})
\end{aligned}
$$

Thus, from equation (1) above,

$$
\begin{aligned}
\frac{d \rho}{d t} & =(1716)(510)\left(1.73 \times 10^{-5}\right)+(0.0345)(1716)(0.0167) \\
& =15.1+0.99=16.1 \mathrm{lb} /\left(\mathrm{ft}^{2}\right)(\mathrm{sec})=\frac{16.1}{2116} \\
& =0.0076 \mathrm{~atm} / \mathrm{sec}
\end{aligned}
$$

2.7 In consistent units,

$$
T=-10+273=263 \mathrm{~K}
$$

Thus,

$$
\begin{aligned}
& \rho=p / R T=\left(1.7 \times 10^{4}\right) /(287)(263) \\
& \rho=0.225 \mathrm{~kg} / \mathrm{m}^{3}
\end{aligned}
$$

2.8
$\rho=p / R T=0.5 \times 10^{5} /(287)(240)=0.726 \mathrm{~kg} / \mathrm{m}^{3}$
$v=1 / \rho=1 / 0.726=1.38 \mathrm{~m}^{3} / \mathrm{kg}$
2.9

$F_{p}=$ Force due to pressure $=\int_{0}^{3} p d x=\int_{0}^{3}(2116-10 x) d x$
$=\left[2116 x-5 x^{2}\right]_{0}^{3}=6303 \mathrm{lb}$ perpendicular to wall.
$F_{\tau}=$ Force due to shear stress $=\int_{0}^{3} \tau d x=\int_{0}^{3} \frac{90}{(x+9)^{\frac{1}{2}}} d x$
$=\left[180(x+9)^{\frac{1}{2}}\right]_{0}^{3}=623.5-540=83.5 \mathrm{lb}$ tangential to wall.


Magnitude of the resultant aerodynamic force $=$

$$
R=\sqrt{(6303)^{2}+(835)^{2}}=6303.6 \mathrm{lb}
$$

$\theta=\operatorname{ArcTan}\left(\frac{83.5}{6303}\right)=0.76^{\circ}$
$2.10 \quad V=\frac{3}{2} V_{\infty} \sin \theta$
Minimum velocity occurs when $\sin \theta=0$, i.e., when $\theta=0^{\circ}$ and $180^{\circ}$.

$$
V_{\min }=0 \text { at } \theta=0^{\circ} \text { and } 180^{\circ} \text {, i.e., at its most forward and rearward points. }
$$

Maximum velocity occurs when $\sin \theta=1$, i.e., when $\theta=90^{\circ}$. Hence,

$$
V_{\max }=\frac{3}{2}(85)(1)=127.5 \mathrm{mph} \text { at } \theta=90^{\circ}
$$

i.e., the entire rim of the sphere in a plane perpendicular to the freestream direction.
2.11 The mass of air displaced is

$$
M=(2.2)(0.002377)=5.23 \times 10^{-3} \text { slug }
$$

The weight of this air is

$$
W_{\mathrm{air}}=\left(5.23 \times 10^{-3}\right)(32.2)=0.168 \mathrm{lb}
$$

This is the lifting force on the balloon due to the outside air. However, the helium inside the balloon has weight, acting in the downward direction. The weight of the helium is less than that of air by the ratio of the molecular weights

$$
W_{H_{c}}=(0.168) \frac{4}{28.8}=0.0233 \mathrm{lb}
$$

Hence, the maximum weight that can be lifted by the balloon is

$$
0.168-0.0233=0.145 \mathrm{lb} .
$$

2.12 Let $p_{3}, \rho_{3}$, and $T_{3}$ denote the conditions at the beginning of combustion, and $p_{4}, \rho_{4}$, and $T_{4}$ denote conditions at the end of combustion. Since the volume is constant, and the mass of the gas is constant, then $p_{4}=\rho_{3}=11.3 \mathrm{~kg} / \mathrm{m}^{3}$. Thus, from the equation of state,

$$
p_{4}=\rho_{4} R T_{4}=(11.3)(287)(4000)=1.3 \times 10^{7} \mathrm{~N} / \mathrm{m}^{2}
$$

or,

$$
p_{4}=\frac{1.3 \times 10^{7}}{1.01 \times 10^{5}}=129 \mathrm{~atm}
$$

2.13 The area of the piston face, where the diameter is $9 \mathrm{~cm}=0.09 \mathrm{~m}$, is

$$
A=\frac{\pi(0.09)^{2}}{4}=6.36 \times 10^{-3} \mathrm{~m}^{2}
$$

(a) The pressure of the gas mixture at the beginning of combustion is

$$
p_{3}=\rho_{3} R T_{3}=11.3(287)(625)=2.02 \times 10^{6} \mathrm{~N} / \mathrm{m}^{2}
$$

The force on the piston is

$$
F_{3}=p_{3} A=\left(2.02 \times 10^{6}\right)\left(6.36 \times 10^{-3}\right)=1.28 \times 10^{4} \mathrm{~N}
$$

Since $4.45 \mathrm{~N}=\mathrm{I} \mathrm{lbf}$,

$$
F_{3}=\frac{1.28 \times 10^{4}}{4.45}=2876 \mathrm{lb}
$$

(b) $\quad p_{4}=\rho_{4} R T_{4}=(11.3)(287)(4000)=1.3 \times 10^{7} \mathrm{~N} / \mathrm{m}^{2}$

The force on the piston is

$$
\begin{aligned}
& F_{4}=p_{4} A=\left(1 / 3 \times 10^{7}\right)\left(6.36 \times 10^{-3}\right)=8.27 \times 10^{4} \mathrm{~N} \\
& F_{4}=\frac{8.27 \times 10^{4}}{4.45}=18,579 \mathrm{lb}
\end{aligned}
$$

2.14 Let $p_{3}$ and $T_{3}$ denote conditions at the inlet to the combustor, and $T_{4}$ denote the temperature at the exit. Note: $p_{3}=p_{4}=4 \times 10^{6} \mathrm{~N} / \mathrm{m}^{2}$
(a) $\rho_{3}=\frac{p_{3}}{R T_{3}}=\frac{4 \times 10^{6}}{(287)(900)}=15.49 \mathrm{~kg} / \mathrm{m}^{3}$
(b) $\rho_{4}=\frac{p_{4}}{R T_{4}}=\frac{4 \times 10^{6}}{(287)(1500)}=9.29 \mathrm{~kg} / \mathrm{m}^{3}$
2.151 mile $=5280 \mathrm{ft}$, and 1 hour $=3600 \mathrm{sec}$.

So:

$$
\left(60 \frac{\text { miles }}{\text { hour }}\right)\left(\frac{5280 \mathrm{ft}}{\text { mile }}\right)\left(\frac{1 \text { hour }}{3600 \mathrm{sec}}\right)=88 \mathrm{ft} / \mathrm{sec} .
$$

A very useful conversion to remember is that

$$
60 \mathrm{mph}=88 \mathrm{ft} / \mathrm{sec}
$$

also, $\quad 1 \mathrm{ft}=0.3048 \mathrm{~m}$

$$
\left(88 \frac{\mathrm{ft}}{\mathrm{sec}}\right)\left(\frac{0.3048 \mathrm{~m}}{1 \mathrm{ft}}\right)=26.82 \frac{\mathrm{~m}}{\mathrm{sec}}
$$

Thus

$$
88 \frac{\mathrm{ft}}{\mathrm{sec}}=26.82 \frac{\mathrm{~m}}{\mathrm{sec}}
$$

$2.16692 \frac{\mathrm{miles}}{\text { hour }}\left(\frac{88 \mathrm{ft} / \mathrm{sec}}{60 \mathrm{mph}}\right)=1015 \mathrm{ft} / \mathrm{sec}$
$692 \frac{\text { miles }}{\text { hour }}\left(\frac{26.82 \mathrm{~m} / \mathrm{sec}}{60 \mathrm{mph}}\right)=309.3 \mathrm{~m} / \mathrm{sec}$
2.17 On the front face

$$
F_{f}=p_{f} A=\left(1.0715 \times 10^{5}\right)(2)=2.143 \times 10^{5} \mathrm{~N}
$$

On the back face

$$
F_{b}=p_{b} A=\left(1.01 \times 10^{5}\right)(2)=2.02 \times 10^{5} \mathrm{~N}
$$

The net force on the plate is

$$
F=F_{f}-F_{b}=(2.143-2.02) \times 10^{5}=0.123 \times 10^{5} \mathrm{~N}
$$

From Appendix C,

$$
1 l b_{f}=4.448 \mathrm{~N} .
$$

So,

$$
F=\frac{0.123 \times 10^{5}}{4.448}=2765 \mathrm{lb}
$$

This force acts in the same direction as the flow (i.e., it is aerodynamic drag.)
2.18 Wing loading $=\frac{W}{s}=\frac{10,100}{233}=43.35 \mathrm{lb} / \mathrm{ft}^{2}$

In SI units:

$$
\begin{aligned}
& \frac{W}{s}=\left(43.35 \frac{\mathrm{lb}}{\mathrm{ft}^{2}}\right)\left(\frac{4.448 \mathrm{~N}}{1 \mathrm{lb}}\right)\left(\frac{1 \mathrm{ft}}{0.3048 \mathrm{~m}}\right)^{2} \\
& \frac{W}{s}=2075.5 \frac{\mathrm{~N}}{\mathrm{~m}^{2}}
\end{aligned}
$$

In terms of kilogram force,

$$
\frac{W}{s}=\left(2075.5 \frac{\mathrm{~N}}{\mathrm{~m}^{2}}\right)\left(\frac{1 \mathrm{k}_{f}}{9.8 \mathrm{~N}}\right)=211.8 \frac{\mathrm{~kg}_{f}}{\mathrm{~m}^{2}}
$$

$2.19 \quad V=\left(437 \frac{\text { miles }}{\mathrm{hr}}\right)\left(\frac{5280 \mathrm{ft}}{\text { mile }}\right)\left(\frac{0.3048 \mathrm{~m}}{1 \mathrm{ft}}\right)=7.033 \times 10^{5} \frac{\mathrm{~m}}{\mathrm{hr}}=703.3 \frac{\mathrm{~km}}{\mathrm{hr}}$
Altitude $=(25,000 \mathrm{ft})\left(\frac{0.3048 \mathrm{~m}}{1 \mathrm{ft}}\right)=7620 \mathrm{~m}=7.62 \mathrm{~km}$
$2.20 \quad V=\left(26,000 \frac{\mathrm{ft}}{\mathrm{sec}}\right)\left(\frac{0.3048 \mathrm{~m}}{1 \mathrm{ft}}\right)=7.925 \times 10^{3} \frac{\mathrm{~m}}{\mathrm{sec}}=7.925 \frac{\mathrm{~km}}{\mathrm{sec}}$
2.21 From Fig. 2.16,
length of fuselage $=33 \mathrm{ft}, 4.125$ inches $=33.34 \mathrm{ft}$

$$
=33.34 \mathrm{ft}\left(\frac{0.3048 \mathrm{~m}}{\mathrm{ft}}\right)=10.16 \mathrm{~m}
$$

wing span $=40 \mathrm{ft}, 11.726$ inches $=40.98 \mathrm{ft}$

$$
=40.98 \mathrm{ft}\left(\frac{0.3048 \mathrm{~m}}{\mathrm{ft}}\right)=12.49 \mathrm{~m}
$$

2.22 (a) From App. C 1 ft. $=0.3048 \mathrm{~m}$.

Thus,

$$
354,200 \mathrm{ft}=(354,000)(0.3048)=107,960 \mathrm{~m}=107.96 \mathrm{~km}
$$

(b) From Example 2.6: $60 \mathrm{mph}=26.82 \mathrm{~m} / \mathrm{sec}$

Thus,

$$
4520 \frac{\text { miles }}{\mathrm{hr}}=4520 \frac{\text { miles }}{\mathrm{hr}}=\frac{\left(26.82 \frac{\mathrm{~m}}{\mathrm{sec}}\right)}{60\left(\frac{\mathrm{mi}}{\mathrm{hr}}\right)} 2020.4 \mathrm{~m} / \mathrm{sec}
$$

2.23

$$
m=\frac{34,000 \mathrm{lb}}{32.2 \mathrm{lb} / \text { slug }}=1055.9 \mathrm{slug}
$$

From Newton's $2^{\text {nd }}$ Law

$$
\begin{aligned}
& F=m a \\
& a=\frac{F}{M}=\frac{57,000}{1055.9}=53.98 \mathrm{ft} / \mathrm{sec}^{2}
\end{aligned}
$$

2.24

$$
\# \text { of } g \prime s=\frac{53.98}{32.2}=1.68
$$

2.25 From Appendix C , one pound of force equals 4.448 N . Thus, the thrust of the Rolls-Royce Trent engine in pounds is

$$
T=\frac{373.7 \times 10^{3} \mathrm{~N}}{4.448 \mathrm{~N} / \mathrm{lb}}=84,015 \mathrm{lb}
$$

### 2.26

(a) $\quad F=(690,000)(9.8)=6.762 \times 10^{6} \mathrm{~N}$
(b) $F=6.762 \times 10^{6} / 4.448=1.5 \times 10^{6} \mathrm{lb}$

## Chapter 3

3.1 An examination of the standard temperature distribution through the atmosphere given in Figure 3.3 of the text shows that both 12 km and 18 km are in the same constant temperature region. Hence, the equations that apply are Eqs. (3.9) and (3.10) in the text. Since we are in the same isothermal region with therefore the same base values of $p$ and $\rho$, these equations can be written as

$$
\frac{\rho_{2}}{\rho_{1}}=\frac{p_{2}}{p_{1}}=e^{-\left(g_{0} / R T\right)\left(h_{2}-h_{1}\right)}
$$

where points 1 and 2 are any two arbitrary points in the region. Hence, with $g_{0}=9.8 \mathrm{~m} / \mathrm{sec}^{2}$ and $R=287$ joule $/ \mathrm{kgK}$, and letting points 1 and 2 correspond to 12 km and 18 km altitudes, respectively,

$$
\frac{\rho_{2}}{\rho_{1}}=\frac{p_{2}}{p_{1}}=e^{\frac{9.8}{(287)(216.66)}(6000)}=0.3884
$$

Hence:

$$
\begin{aligned}
& p_{2}=(0.3884)\left(1.9399 \times 10^{4}\right)=7.53 \times 10^{3} \mathrm{~N} / \mathrm{m}^{2} \\
& \rho_{2}=(0.3884)\left(3.1194 \times 10^{-1}\right)=0.121 \mathrm{~kg} / \mathrm{m}^{3}
\end{aligned}
$$

and, of course,

$$
T_{2}=216.66 \mathrm{~K}
$$

These answers check the results listed in Appendix A of the text within round-off error.
3.2 From Appendix A of the text, we see immediately that $p=2.65 \times 10^{4} \mathrm{~N} / \mathrm{m}^{2}$ corresponds to $10,000 \mathrm{~m}$, or 10 km , in the standard atmosphere. Hence,

$$
\text { pressure altitude = } 10 \mathrm{~km}
$$

The outside air density is

$$
\rho=\frac{p}{R T}=\frac{2.65 \times 10^{4}}{(287)(220)}=0.419 \mathrm{~kg} / \mathrm{m}^{3}
$$

From Appendix A, this value of $\rho$ corresponds to 9.88 km in the standard atmosphere.
Hence,
density altitude $=9.88 \mathrm{~km}$
3.3 At $35,000 \mathrm{ft}$, from Appendix B , we find that $p=4.99 \times 10^{2}=499 \mathrm{lb} / \mathrm{ft}^{2}$.
3.4 From Appendix B in the text,
$33,500 \mathrm{ft}$ corresponds to $p=535.89 \mathrm{lb} / \mathrm{ft}^{2}$
$32,000 \mathrm{ft}$ corresponds to $\rho=8.2704 \times 10^{-4}$ slug $/ \mathrm{ft}^{3}$
Hence,

$$
T=\frac{p}{\rho R}=\frac{535.89}{\left(8.2704 \times 10^{-4}\right)(1716)}=378 R
$$

$3.5 \quad \frac{\left|h-h_{G}\right|}{h}=0.02=\left|1-\frac{h_{G}}{h}\right|$
From Eq. (3.6), the above equation becomes

$$
\begin{aligned}
& \left|1-\left(\frac{r+h_{G}}{r}\right)\right|=\left|1-1-\frac{h_{G}}{r}\right|=0.02 \\
& h_{G}=0.02 r=0.02\left(6.357 \times 10^{6}\right) \\
& h_{G}=1.27 \times 10^{5} \mathrm{~m}=127 \mathrm{~km}
\end{aligned}
$$

$3.6 T=15-0.0065 h=15-0.0065(5000)=-17.5^{\circ} \mathrm{C}=255.5^{\circ} \mathrm{K}$

$$
a=\frac{d T}{d h}=-0.0065
$$

From Eq. (3.12)

$$
\begin{aligned}
\frac{p}{p_{1}} & =\left(\frac{T}{T_{1}}\right)^{-g_{0} / a R}=\left(\frac{255.5}{288}\right)^{-(2.8) /(-0.0065)(287)}=0.533 \\
p & =0.533 p_{1}=0.533\left(1.01 \times 10^{5}\right)=5.38 \times 10^{4} \mathrm{~N} / \mathrm{m}^{2}
\end{aligned}
$$

$3.7 \quad \ln \frac{p}{p_{1}}=-\frac{g}{R T}\left(h-h_{1}\right)$

$$
h-h_{1}=-\frac{1}{g} R T \ln \frac{p}{p_{1}}=-\frac{1}{24.9}(4157)(150) \ln 0.5
$$

Letting $h_{1}=0$ (the surface)

$$
\mathrm{h}=17,358 \mathrm{~m}=17.358 \mathrm{~km}
$$

3.8 A standard altitude of $25,000 \mathrm{ft}$ falls within the first gradient region in the standard atmosphere. Hence, the variation of pressure and temperature are given by:

$$
\begin{equation*}
\frac{p}{p_{1}}=\left(\frac{T}{T_{1}}\right)^{-\frac{g}{a R}} \tag{1}
\end{equation*}
$$

and

$$
\begin{equation*}
T=T_{1}+a\left(h-h_{1}\right) \tag{2}
\end{equation*}
$$

Differentiating Eq. (1) with respect to time:

$$
\begin{equation*}
\frac{1}{p_{1}} \frac{d p}{d t}=\left(\frac{1}{T_{1}}\right)^{-\frac{g}{a R}}\left(-\frac{g}{A R}\right) T^{\left(-\frac{g}{a R}-1\right)} \frac{d T}{d t} \tag{3}
\end{equation*}
$$

Differentiating Eq. (2) with respect to time:

$$
\begin{equation*}
\frac{d T}{d t}=a \frac{d h}{d t} \tag{4}
\end{equation*}
$$

Substitute Eq. (4) into (3)

$$
\begin{equation*}
\left.\frac{d p}{d t}=-p_{1}\left(T_{1}\right)^{\frac{g}{a R}}\left(\frac{g}{R}\right) T^{-\left(\frac{g}{a R}+1\right.}\right) \frac{d h}{d t} \tag{5}
\end{equation*}
$$

In Eq. (5), $d h / d t$ is the rate-of-climb, given by $d h / d t=500 \mathrm{ft} / \mathrm{sec}$. Also, in the first gradient region, the lapse rate can be calculated from the tabulations in Appendix B. For example, take 0 ft and $10,000 \mathrm{ft}$, we find

$$
a=\frac{T_{2}-T_{1}}{h_{2}-h_{1}}=\frac{483.04-518.69}{10,000-0}=-0.00357 \frac{{ }^{\circ} R}{\mathrm{ft}}
$$

Also from Appendix $\mathrm{B}, p_{1}=2116.2 \mathrm{lb} / \mathrm{ft}^{2}$ at sea level, and $T=429.64^{\circ} \mathrm{R}$ at $25,000 \mathrm{ft}$.
Thus,

$$
\frac{g}{a R}=\frac{32.2}{(-0.00357)(1716)}=-5.256
$$

Hence, from Eq. (5)

$$
\begin{aligned}
\frac{d p}{d t} & =-(2116.2)(518.69)^{-5.256}\left(\frac{32.2}{1716}\right)(429.64)^{4.256}(500) \\
\frac{d p}{d t} & =-17.17 \frac{\mathrm{lb}}{\mathrm{ft}^{2} \mathrm{sec}}
\end{aligned}
$$

3.9 From the hydrostatic equation, Eq. (3.2) or (3.3),

$$
\begin{aligned}
d p & =-\rho g_{0} d h \\
\text { or } \quad \frac{d p}{d t} & =-\rho g_{0} \frac{d h}{d t}
\end{aligned}
$$

The upward speed of the elevator is $d h / d t$, which is

$$
\frac{d p}{d t}=\frac{d p / d t}{-\rho g_{0}}
$$

At sea level, $\rho=1.225 \mathrm{~kg} / \mathrm{m}^{3}$. Also, a one-percent change in pressure per minute starting from sea level is

$$
\frac{d p}{d t}=-\left(1.01 \times 10^{5}\right)(0.01)=-1.01 \times 10^{3} \mathrm{~N} / \mathrm{m}^{2} \text { per minute }
$$

Hence,

$$
\frac{d h}{d t}=\frac{-1.01 \times 10^{3}}{(1.225)(9.8)}=84.1 \text { meter per minute }
$$

### 3.10 From Appendix B:

At $35,500 \mathrm{ft}: p=535.89 \mathrm{lb} / \mathrm{ft}^{2}$
At $34,000 \mathrm{ft}: p=523.47 \mathrm{lb} / \mathrm{ft}^{2}$
For a pressure of $530 \mathrm{lb} / \mathrm{ft}^{2}$, the pressure altitude is

$$
33,500+500\left(\frac{535.89-530}{535.89-523.47}\right)=33737 \mathrm{ft}
$$

The density at the altitude at which the airplane is flying is

$$
\rho=\frac{\rho}{R T}=\frac{530}{(1716)(390)}=7.919 \times 10^{-4} \text { slug } / \mathrm{ft}^{3}
$$

From Appendix B:

$$
\begin{aligned}
& \text { At } 33,000 \mathrm{ft}: \rho=7.9656 \times 10^{-4} \text { slug } / \mathrm{ft}^{3} \\
& \text { At } 33,500 \mathrm{ft}: \rho=7.8165 \times 10^{-4} \text { slug } / \mathrm{ft}^{3}
\end{aligned}
$$

Hence, the density altitude is
$33,000+500\left(\frac{7.9656-7.919}{7.9656-7.8165}\right)=33,156 \mathrm{ft}$
3.11 Let $\ell$ be the length of one wall of the tank, $\ell=30 \mathrm{ft}$. Let $d$ be the depth of the pool, $d=10 \mathrm{ft}$. At the water surface, the pressure is atmospheric pressure, $p_{\mathrm{a}}$. The water pressure increases with increasing depth; the pressure as a function of distance below the surface, $h$, is given by the hydrostatic equation

$$
\begin{equation*}
d p=\rho g d h \tag{1}
\end{equation*}
$$

Note: The hydrostatic equation given by Eq. (3.2) in the text has a minus sign because $h_{G}$ is measured positive in the upward direction. In Eq. (1), $h$ is measured positive in the downward direction, with $h=0$ at the surface of the water. Hence, no minus sign appears in Eq. (1); as $h$ increases (as we go deeper into the water), $p$ increases. Eq. (1) is consistent with this fact. Integrating Eq. (1) from $h=0$ where $p=p_{a}$ to some local depth $h$ where the pressure is $p$, and noting that $\rho$ is constant for water, we have

$$
\int_{p_{a}}^{\rho} d p=\rho g \int_{0}^{h} d h
$$

or, $\quad p-p_{a}=\rho g h$
or, $\quad p=\rho g h+p_{a}$
Eq. (2) gives the water pressure exerted on the wall at an arbitrary depth $h$.
Consider an elementary small sliver of wall surface of length $\ell$ and height $d h$.
The water force on this sliver of area is

$$
d F=p \ell d h
$$

Total force, F , on the wall is

$$
\begin{equation*}
F=\int_{0}^{F} d F=\int_{0}^{d} p \ell d h \tag{3}
\end{equation*}
$$

where $p$ is given by Eq. (2). Inserting Eq. (2) into (3),

$$
\begin{align*}
F & =\int_{0}^{d}\left(\rho g h+p_{a}\right) \ell d h \\
\text { or, } \quad F & =\rho g \ell\left(\frac{d^{2}}{2}\right)+p_{a} \ell d \tag{4}
\end{align*}
$$

In Eq. (4), the product $\rho g$ is the specific weight (weight per unit volume) of water; $\rho g=62.4 \mathrm{lb} / \mathrm{ft}^{3}$. From Eq. (4),

$$
\begin{aligned}
& F=(62.4)(30) \frac{(10)^{2}}{2}+(2116)(30)(10) \\
& F=93,600+634,800=728,400 \mathrm{lb}
\end{aligned}
$$

Note: This force is the combined effect of the force due to the weight of the water, $93,600 \mathrm{lb}$, and the force due to atmospheric pressure transmitted through the water, 634,800 lb. In this example, the latter is the larger contribution to the force on the wall. If the wall were freestanding with atmospheric pressure exerted on the opposite side, then the net force exerted on the wall would be that due to the weight of the water only, i.e., $93,600 \mathrm{lb}$. In tons, the force on the side of the wall in contact with the water is

$$
F=\frac{728,400}{2000}=364.2 \text { tons }
$$

In the case of a freestanding wall, the net force, that due only to the weight of the water, is

$$
F=\frac{93,600}{2000}=46.8 \mathrm{tons}
$$

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3.12 For the exponential atmosphere model,

$$
\begin{aligned}
& \frac{\rho}{\rho_{0}}=e^{-g_{0} h /(R T)} \\
& \frac{\rho}{\rho_{0}}=e^{-(9.8)(45,000) /(287)(240)}=e^{-6.402}
\end{aligned}
$$

Hence,

$$
\rho=\rho_{0} e^{-6.402}=(1225)\left(1.6575 \times 10^{-3}\right)=2.03 \times 10^{-3} \mathrm{~kg} / \mathrm{m}^{3}
$$

From the standard atmosphere, at $45 \mathrm{~km}, p=2.02 \times 10^{-3} \mathrm{~kg} / \mathrm{m}^{3}$. The exponential atmosphere model gives a remarkably accurate value for the density at 45 km when a value of 240 K is used for the temperature.
3.13 At 3 km , from App. A, $T=268.67 \mathrm{~K}, p=7.0121 \times 10^{4} \mathrm{~N} / \mathrm{m}^{2}$, and $\rho=0.90926 \mathrm{~kg} / \mathrm{m}^{3}$. At 3.1 km , from App. A, $T=268.02 \mathrm{~K}, p=6.9235 \times 10^{4} \mathrm{~N} / \mathrm{m}^{2}$, and $\rho=0.89994 \mathrm{~kg} / \mathrm{m}^{3}$. At $h=3.035 \mathrm{~km}$, using linear interpolation:

$$
\begin{aligned}
& T=268.67-(268.67-268.02)\left(\frac{0.035}{0.1}\right)=268.44 \mathrm{~K} \\
& p=7.0121 \times 10^{4}-\left(7.0121 \times 10^{4}-6.9235 \times 10^{4}\right)\left(\frac{0.035}{0.1}\right) \\
&=6.98099 \times 10^{4} \frac{\mathrm{~N}}{\mathrm{~m}^{2}} \\
& \rho=0.09026-(0.90926-0.89994)\left(\frac{0.035}{0.1}\right)=0.906 \frac{\mathrm{~kg}}{\mathrm{~m}^{3}}
\end{aligned}
$$

